

A Comparison between the Perturbed-Chain Statistical Associating Fluid Theory Equation of State and Machine Learning Modeling Approaches in Asphaltene Onset Pressure and Bubble Point Pressure Prediction during Gas Injection

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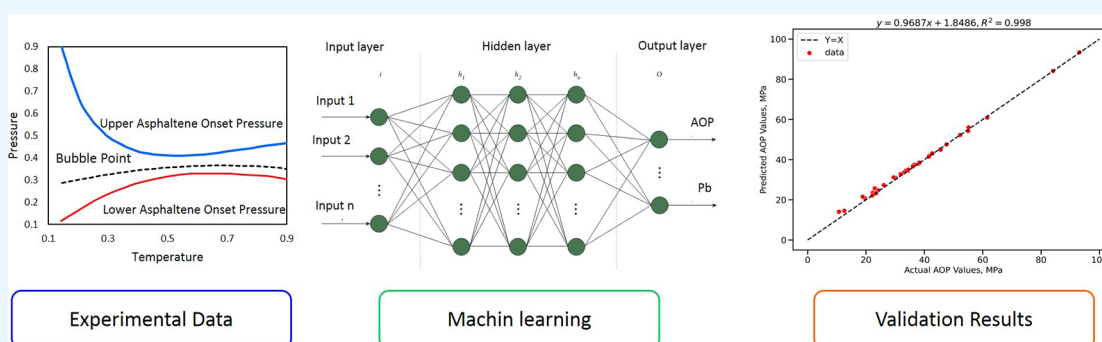
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ABSTRACT: Predicting asphaltene onset pressure (AOP) and bubble point pressure (Pb) is essential for optimization of gas injection for enhanced oil recovery. Pressure-Volume-Temperature or PVT studies along with equations of state (EoSs) are widely used to predict AOP and Pb. However, PVT experiments are costly and time-consuming. The perturbed-chain statistical associating fluid theory or PC-SAFT is a sophisticated EoS used for prediction of the AOP and Pb. However, this method is computationally complex and has high data requirements. Hence, developing precise and reliable smart models for prediction of the AOP and Pb is inevitable. In this paper, we used machine learning (ML) methods to develop predictive tools for the estimation of the AOP and Pb using experimental data (AOP data set: 170 samples; Pb data set: 146 samples). Extra trees (ET), support vector machine (SVM), decision tree, and k-nearest neighbors ML methods were used. Reservoir temperature, reservoir pressure, SARA fraction, API gravity, gas–oil ratio, fluid molecular weight, monophasic composition, and composition of gas injection are considered as input data. The ET (R^2 : 0.793, RMSE: 7.5) and the SVM models (R^2 : 0.988, RMSE: 0.76) attained more reliable results for estimation of the AOP and Pb, respectively. Generally, the accuracy of the PC-SAFT model is higher than that of the AI/ML models. However, our results confirm that the AI/ML approach is an acceptable alternative for the PC-SAFT model when we face lack of data and/or complex mathematical equations. The developed smart models are accurate and fast and produce reliable results with lower data requirements.

1. INTRODUCTION

Flow assurance issues are well known in the petroleum industry spanning from upstream to downstream. Asphaltene precipitation and deposition is one of the main issues in this industry that occurs due to changes in temperature, pressure, and oil composition. Asphaltenes are crude oil's polar components with a complex molecular structure and different molecular weights (from 500 to 10,000 g/mol).¹ Based on their solubility, asphaltenes dissolve in aromatic solvents (e.g., toluene, naphtha, and benzene) and are nonsoluble in *n*-alkanes. Clogging porous media, well plugging, permeability reduction, and wettability alteration are some of the significant

consequences of asphaltene precipitation, which occur during different oil production stages.^{2,3}

Injection of gas (e.g., nitrogen, carbon dioxide, and natural gas) during EOR operations is one of the widely used and effective ways to recover residual oil after primary and

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secondary oil recovery methods. The injection gases can enhance the flow of crude oil via reducing its viscosity.³ However, a combination of oil and gas changes the composition and behavior of reservoir fluids and provides a favorable situation for asphaltene precipitation.² Hence, estimation of asphaltene onset pressure (AOP) is vital to minimize asphaltene precipitation and the related problems.

Accurate estimation of the AOP in various stages of oil production can help to have a correct long-term plan from the economic viewpoint. Various research studies (experimental and modeling) addressed the main problems in the petroleum industry in the field of kinetics, AOP, and the amount of asphaltene precipitation/deposition. Light scattering, high-pressure microscopy, gravimetric precipitation, quartz crystal resonators, and isobaric filtration techniques are the main experimental methods that are used for estimating the AOP.^{4–6} Moreover, there are many modeling approaches to assess the kinetic and asphaltene thermodynamic states under various thermodynamic conditions.

Thermodynamic solubility techniques and colloidal-based models are two main groups of modeling asphaltene precipitation considering the assumption about the stability state of asphaltene in crude oil.^{7,8} In the colloidal approach, asphaltene is assumed as a solid particle peptized by resin in the solution. Based on this theory, the asphaltene aggregation will be formed when resins separate from asphaltene molecules. In addition, the asphaltene precipitation assumes an irreversible phenomenon in the colloidal theory. In the solubility model, asphaltenes are considered soluble particles in crude oil. In addition, this theory assumes that the dispersion force between asphaltene and other fractions in crude oil is responsible for asphaltene aggregation, not polar interaction. Asphaltenes will precipitate by decreasing their concentration below a critical solubility value.⁹

The solubility models fall into two main categories of regular solution theory and equation of state (EOS). The first group considers that the system is made of crude oil and asphaltene. The regular solution theory is used generally for polymer solution and studies based on Flory–Huggins,¹⁰ Scatchard–Hildebrand,¹¹ and Scott–Magat theories.¹² EOS theory is considered as a mixture of pseudo- and pure components.⁹ Cubic, Cubic Plus Association, statistical associating fluid theory (SAFT), Peng Robinson, and Soave–Redlich–Kwong are placed in EOS theory.^{7,9} Among these, the SAFT model is the most reliable model to investigate thermodynamic behavior of asphaltene as a complex mixture because of the association effect, and nonspherical chains are considered in this theory.¹³ The perturbed-chain form statistical associating fluid theory (PC-SAFT) is the most useful SAFT model to predict the stability of asphaltene during EOR methods (e.g., gas injection), drop of pressure, and PVT studies that were proposed and developed by Gross and Sadowski.¹⁴

Several research studies applied the SAFT model to simulate asphaltene stability in the system. Ting et al.¹³ used the SAFT model for this aim. They classified crude oil components into three gas pseudocomponents and three liquid phases. Their model successfully predicted the instability and bubble point of asphaltene compared to experimental data. Following that, Panuganti et al.¹⁵ improved the parameters used by Ting et al.¹³ and considered the effect of gas injection on the stability of asphaltene in the system. They reported that considering light components as a separate group, the model's accuracy in predicting bubble point and AOP was enhanced. Gonzalez et

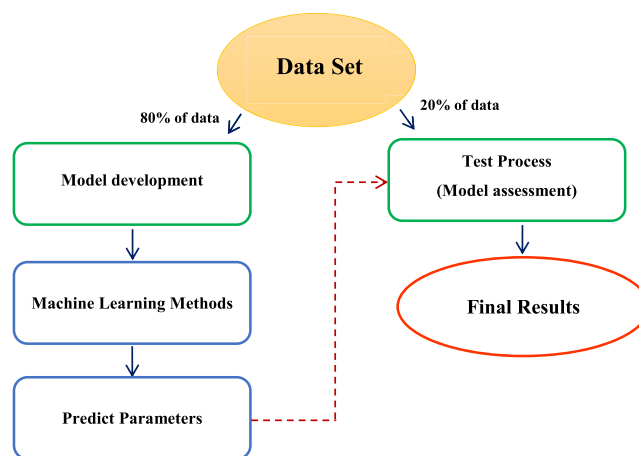


Figure 1. Flowchart showing the AI/ML modeling workflow used in this research work for prediction of AOP and Pb.

Table 1. Number and Percentage of Missing Values for Each Parameter

parameter	number of missing data	percentage of missing data (%)
reservoir pressure	50	29.41
reservoir temperature	31	18.23
API gravity	0	0
GOR	14	8.23
fluid molecular weight	5	2.94
oil-N ₂	0	0
oil-CO ₂	0	0
oil-H ₂ S	0	0
oil-C ₁	0	0
oil-C ₂	0	0
oil-C ₃	0	0
oil-i-n-C ₄	0	0
oil-i-n-C ₅	0	0
oil-C ₆	0	0
saturates	5	2.94
aromatics	5	2.94
resins	5	2.94
asphaltenes	5	2.94
N ₂	0	0
CO ₂	0	0
H ₂ S	0	0
C ₁	0	0
C ₂	0	0
C ₃	0	0
C ₄₊	0	0
temperature	0	0
AOP	0	0
bubble point pressure (Pb)	24	14.12

al.¹⁶ investigated the effect of signal component gas (e.g., CH₄, N₂, and CO₂) on asphaltene stability. In another study, they used the PC-SAFT model for accessing asphaltene precipitation in the presence of oil-based mud and considering the gas-oil-ratio (GOR).¹⁷ In addition to results of these studies, the thermodynamic modeling approach is costly, time-consuming, usually complicated, and unsuitable for nonlinear systems. Hence, it is necessary to use methods that cover this limitation because asphaltene precipitation and deposition is a nonlinear and complex process.

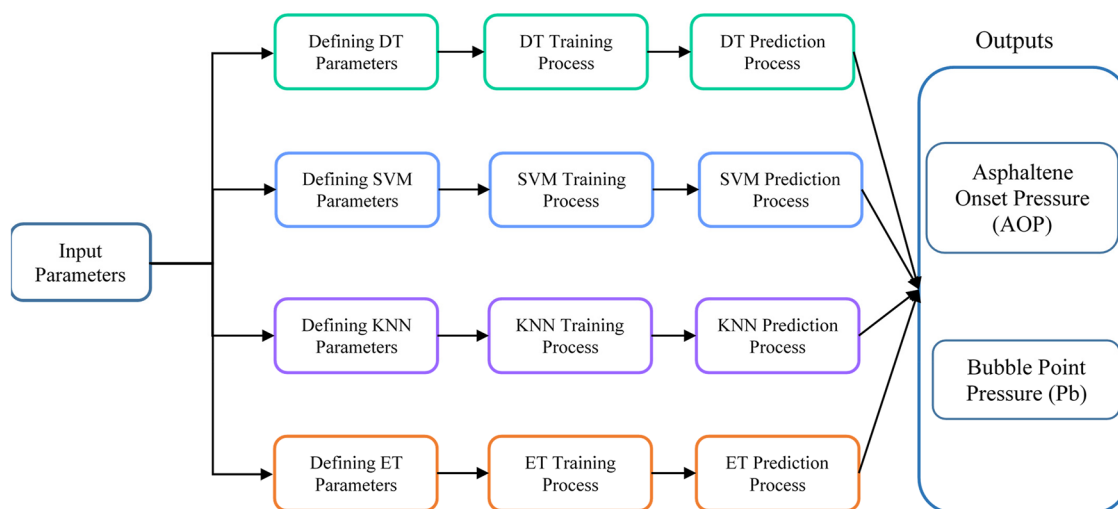


Figure 2. Overall schematic flowchart for the used models.

Smart computational tools are considered practical methods to predict the target parameters without knowing the mathematical/empirical relations and experimental procedure. Machine learning (ML) methods, which are data-driven models, attempt to find a function relating predictors to output.¹⁸ This approach provides a suitable method to predict asphaltene thermodynamic behavior as a nonlinear and discontinuous function. Artificial neural network (ANN), least-squares support vector machine (LSSVM), support vector machine (SVM) and support vector regression (SVR), random forest (RF), genetic algorithm (GA), decision tree (DT), and particle swarm optimization (PSO) are the most common AI algorithms used for the prediction of asphaltene stability parameters. Khamsehchi et al.¹⁹ used ANN and adaptive neural fuzzy system models to predict bubble point and AOP during CO₂ injection. Their results are in good agreement with experimental data and confirm validity of their models. Zendejboudi et al.¹⁸ investigated the effect of miscible gas injection during EOR using the PSO-ANN method. Based on their results, temperature has a great impact on minimum miscibility pressure (MMP), and a good agreement between their modeling and experimental results was observed. Kamari et al.²⁰ predicted AOP and oil saturation conditions using the LSSVM. They considered different crude oils mostly from the Middle East with different asphaltene contents. Their results were by comparing the results with experimental data. In another study, LSSVM and a multilayer extreme learning machine were used to estimate asphaltene bubble pressure by Rashidi et al.²¹ They used the GA and PSO to optimize the methods. In addition to the accuracy of their models in predicting bubble point and oil formation volume factor, the results show that gas–oil gravity is an influential factor in bubble point prediction compared with other parameters in their work. Although the results of previous studies are acceptable, the effects of various types of gas injection on AOP and bubble point and the relation between ML results and the PC-SAFT model were not considered in previous research studies reported in the literature.

The main objective of the present study was to use AI/ML methods to estimate bubble point pressure and AOP and compare the results with the PC-SAFT model during gas injection. To the best of our knowledge, this is the first research work where the outcomes of the PC-SAFT model and

AI/ML-based smart models in predicting AOP and bubble point pressure during gas injection are compared.

This paper is organized into four main parts. After Introduction, methodology is described in detail, and data preprocessing, model development, and evaluation with the assessment of statistical error and PC-SAFT EoS are presented. Subsequently, in the Results and Discussion section, modeling results, evaluation of the developed models, and comparisons between the developed smart models with PC-SAFT EoS are discussed. Finally, the main outcomes are highlighted in the Conclusions section.

2. METHODOLOGY

The overall procedure for this modeling research work consists of three steps: data collection, data preprocessing, AI/ML model development, and model evaluation. Each step is discussed in detail in the following subsections.

2.1. Data Collection. Various parameters affect AOP and bubble point pressure during gas injection. These parameters include temperature, pressure, molecular weight of asphaltene, oil composition, oil properties (e.g., API and SARA), GOR, and properties of gas injection. For the purpose of this research work, we used experimental data from the literature where the effect of these parameters on AOP and Pb was investigated by several researchers.^{15,22–43} The collected data were then split into two sets of training data that include 80% of the data set (AOP data set: 136 samples; Pb data set: 116 samples) and testing data that contain 20% (AOP data set: 34 samples; Pb data set: 30 samples) of the total data. Training data are used to train and build AI/ML models, and the testing data are used to evaluate the performance of the developed smart models in estimating the desired parameter. A flowchart illustrating the AI/ML modeling approach for prediction of AOP and Pb is presented in Figure 1.

2.2. Data Preprocessing. **2.2.1. Missing Value Imputation.** The issue of missing values in data-driven modeling approaches is quite common when analyzing real data sets. Missing values can cause a bias in modeling results and affect model performance. Hence, a key step in data modeling is handling missing value. In this research work, AOP and Pb are target variables. Hence, samples with missing values in these two parameters were removed from the database. The number of missing values in each parameter must first be specified to

Table 2. Statistical Information of the Data Set for AOP

parameter	count	unit	range	mean (standard deviation)
reservoir pressure	170	MPa	19.68–137.90	42.01 (16.16)
reservoir temperature	170	K	294.25–418.87	383.13 (21.17)
API gravity	170		20.32–41.69	34.37 (5.84)
GOR	170	m ³ /m ³	38.12–314.78	142.39 (66.02)
fluid molecular weight	170	g/mol	66.69–169.00	108.37 (26.56)
oil-N ₂	170	mol %	0.02–6.59	0.52 (1.06)
oil-CO ₂	170	mol %	0.05–3.68	1.54 (0.97)
oil-H ₂ S	170	mol %	0.00–5.39	0.44 (1.26)
oil-C ₁	170	mol %	12.39–61.00	34.11 (11.52)
oil-C ₂	170	mol %	4.33–12.42	7.59 (2.01)
oil-C ₃	170	mol %	2.48–10.29	6.18 (1.29)
oil-i-n-C ₄	170	mol %	0.00–6.90	4.38 (2.02)
oil-i-n-C ₅	170	mol %	0.00–7.85	3.77 (1.55)
oil-C ₆	170	mol %	0.00–5.48	2.95 (1.48)
saturates	170	wt %	24.80–80.64	61.43 (12.39)
aromatics	170	wt %	11.60–47.81	26.84 (8.22)
resins	170	wt %	1.47–20.60	8.36 (4.57)
asphaltenes	170	wt %	0.10–15.50	2.88 (4.61)
N ₂	170	mol %	0.00–10.00	0.34 (1.61)
CO ₂	170	mol %	0.00–45.00	4.07 (9.62)
H ₂ S	170	mol %	0.00–10.00	0.22 (1.33)
C ₁	170	mol %	0.00–46.52	5.00 (10.09)
C ₂	170	mol %	0.00–7.75	0.78 (1.62)
C ₃	170	mol %	0.00–5.94	0.44 (1.00)
C ₄₊	170	mol %	0.00–5.21	0.29 (0.72)
temperature	170	K	294.26–424.55	362.62 (30.56)
asphaltene onset pressure (AOP)	170	MPa	10.71–135.14	41.26 (19.99)

deal with the remaining missing values. A parameter must be eliminated if the parameter has many missing values or the percentage of missing values exceeds a certain threshold. In this research work, a threshold of 50% was considered according to previous research studies.⁴⁴ As shown in Table 1, no parameter has more than 50% missing values, and therefore, they remain. For the other parameters, the missing values are replaced with the mean value of the parameter which is a widely used method for estimating missing values for numeric features. There are many approaches available for replacing missing values with imputed values, and every single of them has its own pros and cons. A detailed discussion of those methods is beyond the scope of this research work. In this research work, one of the most used and popular methods was used, which is mean (average) imputation as a quick and simple way for handling missing values, and in this case, fair results have been achieved.⁴⁵

2.2.2. Data Normalization. Data normalization was used to normalize the range of variables such that the range of features is the same. This process is commonly performed during the data preprocessing step to handle highly varying magnitudes or values or units. All input parameters have the same effect on the model development process, ultimately. In this research work, the standardization technique was used for data normalization that causes all parameters have a mean and standard deviation of 0 and 1, respectively. A standard scalar is first fitted on the training data and then applied to the testing data.

Table 3. Statistical Information of the Data Set for Pb

parameter	count	unit	range	mean (standard deviation)
reservoir pressure	146	MPa	19.68–89.63	41.46 (12.44)
reservoir temperature	146	K	294.25–418.87	384.09 (22.26)
API gravity	146		20.32–41.69	34.87 (5.99)
GOR	146	m ³ /m ³	56.81–314.78	152.63 (62.20)
fluid molecular weight	146	g/mol	66.69–169.00	103.43 (21.77)
oil-N ₂	146	mol %	0.02–6.59	0.58 (1.13)
oil-CO ₂	146	mol %	0.05–3.68	1.65 (0.99)
oil-H ₂ S	146	mol %	0.00–5.39	0.51 (1.35)
oil-C ₁	146	mol %	16.84–61.00	35.62 (10.06)
oil-C ₂	146	mol %	4.33–12.42	7.78 (1.92)
oil-C ₃	146	mol %	2.48–10.29	6.21 (1.32)
oil-i-n-C ₄	146	mol %	0.00–6.90	4.37 (2.04)
oil-i-n-C ₅	146	mol %	0.00–7.85	3.74 (1.52)
oil-C ₆	146	mol %	0.00–5.48	2.89 (1.43)
saturates	146	wt %	24.80–80.64	61.99 (13.06)
aromatics	146	wt %	11.60–47.81	27.14 (8.58)
resins	146	wt %	1.47–20.60	7.66 (4.32)
asphaltenes	146	wt %	0.10–15.50	2.88 (4.77)
N ₂	146	mol %	0.00–10.00	0.29 (1.63)
CO ₂	146	mol %	0.00–45.00	3.73 (9.35)
H ₂ S	146	mol %	0.00–10.00	0.23 (1.42)
C ₁	146	mol %	0.00–39.46	3.23 (7.47)
C ₂	146	mol %	0.00–6.79	0.45 (1.07)
C ₃	146	mol %	0.00–5.15	0.26 (0.72)
C ₄₊	146	mol %	0.00–5.21	0.23 (0.70)
temperature	146	K	294.26–418.71	362.37 (31.35)
bubble point pressure (Pb)	146	MPa	5.38–43.21	(7.17)

2.2.3. Model Development. Four AI/ML models that have been shown to be effective in data modeling problems were explored in this research work: SVM, KNN, and DT, and ET. A brief description of each AI/ML algorithm is provided in the text below.

2.2.4. Support Vector Machine. The SVM is a powerful tool in regression and classification that was introduced in the 1960s.⁴⁵ The training data are classified in different classes in this method and show a good performance in modeling noisy and high-dimension data. In addition, it is suitable for a database containing a high number of features compared to the number of samples.⁴⁶

2.2.5. k-Nearest Neighbor. The KNN method predicts a class label for a given sample based on its K neighborhoods.⁴⁷ The KNN algorithm works by finding the K examples closest to the query first and then averages the labels of these K examples to estimate the value for the query. We used the KNN algorithm because it is easy to use and effective and robust in data modeling.

2.2.6. Decision Tree. The DT algorithm provides a rule-based model by which the value of a target variable is estimated using decision rules extracted from the data features.⁴⁸ The extracted tree can be considered as a piecewise constant approximation. For the sake of brevity, the DT algorithm is not discussed in detail; such information is provided elsewhere.

2.2.7. Extra Trees. The ET algorithm is a type of ensemble learning technique in which the results of different de-correlated DT algorithms are aggregated to provide output

Table 4. Result of the AI/ML Model Development for Prediction of AOP

model	subset	R^2 (STD)	RMSE (STD)
extra trees	training	1.00 (0.00)	6.72×10^{-14} (0.00)
	testing	0.79 (1.11×10^{-16})	7.51 (1.78×10^{-15})
k-nearest neighbors	training	1.00 (0.00)	0.00 (0.00)
	testing	0.62 (2.22×10^{-16})	10.16 (0.00)
decision tree	training	0.71 (2.22×10^{-16})	1.10×10^1 (1.78×10^{-15})
	testing	0.48 (1.10×10^{-16})	11.87 (1.78×10^{-15})
support vector machine	training	0.97 (3.33×10^{-16})	3.61 (4.44×10^{-16})
	testing	0.76 (2.22×10^{-16})	7.99 (8.88×10^{-16})

Table 5. Result of AI/ML Model Development for Prediction of Pb

model	subset	R^2 (STD)	RMSE (STD)
extra trees	training	1.00 (0.00)	2.97×10^{-14} (0.00)
	testing	0.97 (2.22×10^{-16})	1.71 (6.66×10^{-16})
k-nearest neighbors	training	1.00 (0.00)	0.00 (0.00)
	testing	0.94 (0.00)	1.84 (6.66×10^{-16})
decision tree	training	0.99 (0.00)	2.54×10^{-01} (0.00)
	testing	0.88 (3.33×10^{-16})	2.51 (4.44×10^{-16})
support vector machine	training	0.98 (3.33×10^{-16})	9.36×10^{-01} (1.11×10^{-16})
	testing	0.99 (0.00)	0.79 (0.00)

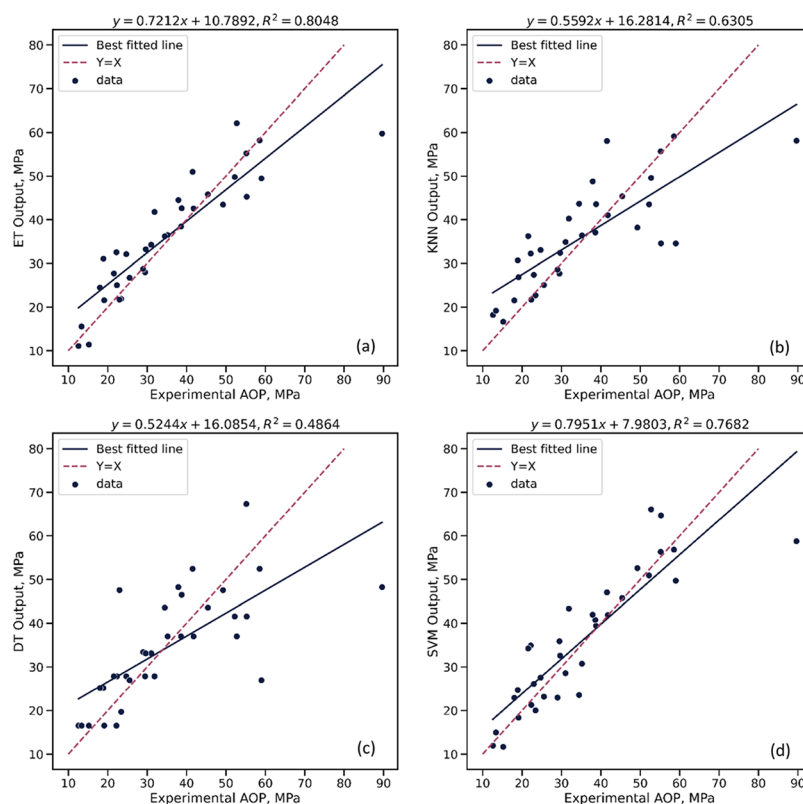
of the model. In the ET method, a group of trees are formed randomly, and their results are aggregated.⁴⁹ The overall flowchart of all models is shown in Figure 2.

Each AI/ML model contains several essential parameters known as hyperparameters that significantly influence the model performance in predicting the output parameter. It

should be emphasized here that hyperparameter optimization must be completed before training the models.^{50,51} In this research work, an automatic hyperparameter optimization framework specifically intended for ML known as Optuna⁵² is used in conjunction with fivefold cross-validation on training data to optimize hyperparameters. To accomplish this, we first determine the maximum and lowest values of each hyperparameter, and then Optuna creates a model by arranging these parameters in a random order. Then performance of the model is evaluated using 5-fold cross-validation. As a result, the model will be constructed in such a way that the root mean square error (RMSE) is minimized. This method is repeated until the RMSE is as low as possible.⁵²

2.2.8. Model Evaluation. We used two well-known indices, namely, coefficient of determination (R^2) and RMSE as defined below to evaluate the performance of the developed models:⁵³

$$R^2 = 1 - \frac{\sum_{i=1}^N (y_i - \hat{y}_i)^2}{\sum_{i=1}^N (y_i - \bar{y})^2} \quad (1)$$

**Figure 3.** Predicted versus experimental values of AOP for (a) ET model, (b) KNN model, (c) DT model, and (d) SVM model in the testing data.

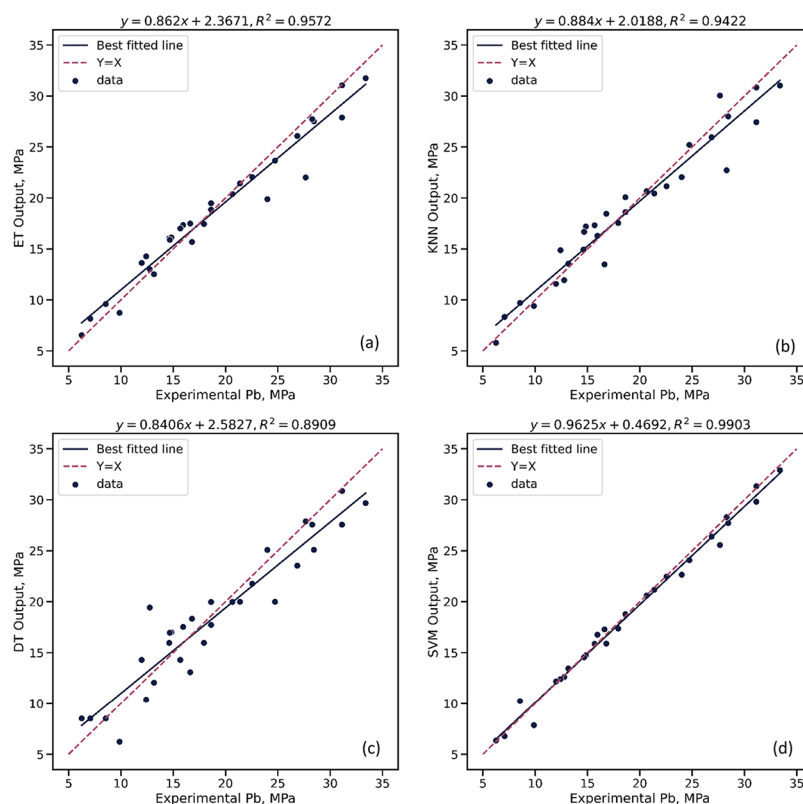


Figure 4. Predicted versus experimental values of Pb for (a) ET model, (b) KNN, (c) DT model, and (d) SVM model in the testing data.

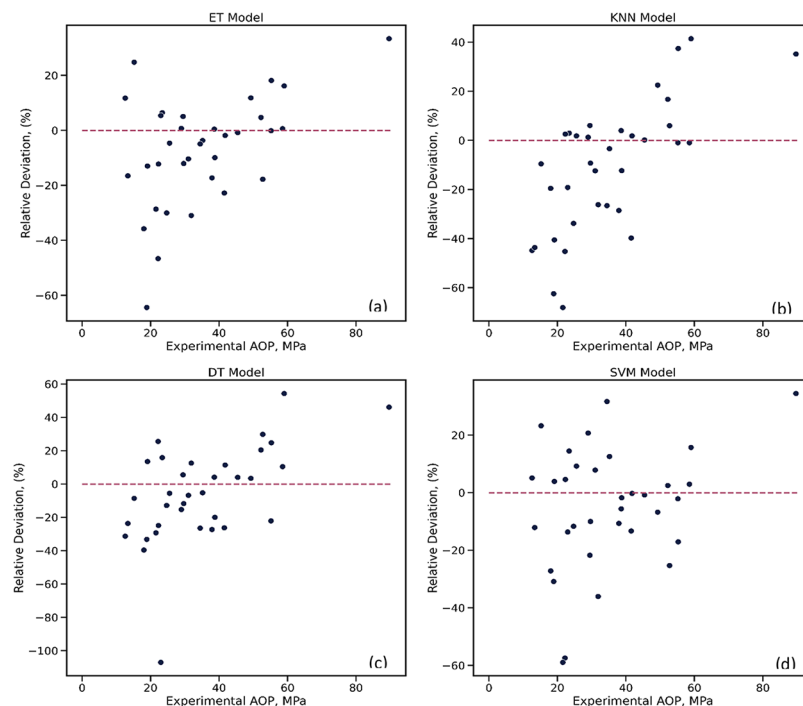


Figure 5. Relative deviation of the model predictions in the testing data set for AOP: (a) ET model, (b) KNN model, (c) DT model, and (d) SVM model.

$$\text{RMSE} = \sqrt{\frac{1}{N} \sum_{i=1}^N (y_i - \hat{y}_i)^2} \quad (2)$$

where \hat{y}_i and \bar{y} are the predicted value for the i th sample and average values, respectively.

It should be emphasized here that we used only the testing data set to assess the performance of the models to avoid bias in estimating these indices that could mislead evaluation of the models. The training data set was used only for training the models.

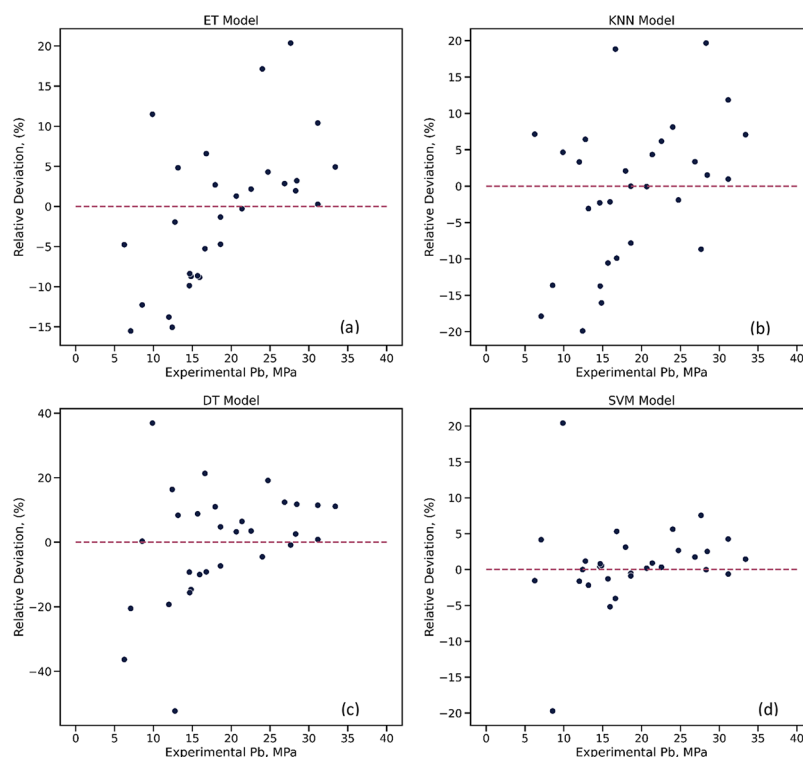


Figure 6. Relative deviation of the model predictions in the testing data set for Pb: (a) ET model, (b) KNN model, (c) DT model, and (d) SVM model.

The reliability of the developed models was further tested using some noisy data samples. Twenty percent of the total data (34 samples for the AOP data set and 29 samples for the Pb data set) was randomly selected. Then, a Gaussian noise (with mean equal to zero and specified standard deviation) was added to each feature in this data set for generating the noisy samples. The standard deviation of the noises was determined based on standard deviation of each feature (2% changes) in the randomly selected data samples for creating feasible values. Finally, the AI/ML model results (e.g., ET, DT, KNN, and SVM) were compared with the PC-SAFT model results to evaluate accuracy of the AI/ML models in this research work.

2.2.9. PC-SAFT Model Characterization. The SAFT model assumes that all molecules are spherical segments with equal sizes. This method considers five main pure components including segment number (m), segment diameter (σ), segment energy (ϵ), associate energy ($\epsilon_{A,B}$), and associate volume ($\kappa_{A,B}$). Associate energy and associate volume are important when self-associating is considered between molecules. The SAFT model is modified by considering dipole–dipole interactions using the Lennard–Jones equation for calculation of the segment contribution. This modified model is called PC-SAFT.⁵⁴ The PC-SAFT model successfully can be used to model chain molecules like hydrocarbon solution because of considering dispersion interaction between them. Helmholtz free energy is used to calculate thermodynamic properties. Hence, the PC-SAFT model uses dimensionless residual Helmholtz energy to estimate thermodynamic properties. The following equation shows this energy with considering different types of interactions.^{31,55}

$$\tilde{a}^{\text{res}} = \tilde{a}^{\text{hc}} + \tilde{a}^{\text{disp}} + \tilde{a}^{\text{assoc}} \quad (3)$$

where dispersion forces (\tilde{a}^{disp}) show the interaction among nonpolar and weakly polar molecules. The hydrogen interaction and donor-acceptor forces show using association contribution (\tilde{a}^{assoc}), while hard-chain formation and sphere contribution demonstrate using hard-chain contribution (\tilde{a}^{hc}). These parameters and residual Helmholtz energy (\tilde{a}^{res}) are calculated using eqs 4–6 as follows:

$$\left(\frac{\tilde{a}^{\text{hc}}}{RT}\right) = \left(\frac{a_0^{\text{hs}}}{RT}\right) \sum_i X_i m_i + \frac{a^{\text{assoc}}}{RT} \quad (4)$$

$$\left(\frac{\tilde{a}^{\text{assoc}}}{RT}\right) = \sum_i X_i (1 - m_i) \ln g_{ii}(d_{ii}) \quad (5)$$

$$\left(\frac{\tilde{a}_0^{\text{hs}}}{RT}\right) = \frac{1}{\zeta_0} \left[\frac{3\zeta_1\zeta_2}{(1-\zeta_3)} + \frac{\zeta_2^3}{\zeta_3(1-\zeta_3)^2} + \left(\frac{\zeta_2^3}{\zeta_3^2} - \zeta_0 \right) \ln(1-\zeta_3) \right] \quad (6)$$

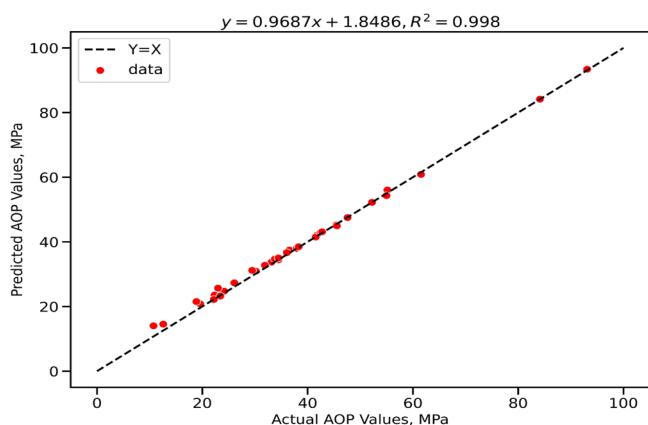
$$d_{ii} = \sigma_i \left[1 - 0.12 \exp\left(-\frac{3\epsilon_{ii}}{k_{\text{BT}}T}\right) \right] \quad (7)$$

$$\tilde{a}^{\text{res}} = \frac{A^{\text{res}}}{NkT} \quad (8)$$

where A^{res} denotes the residual Helmholtz free energy. N , k , and T are the number of molecules, Boltzmann constant, and absolute temperature, respectively. d , x , and m_i are the sphere diameter, molar fraction of a chain component, and number of segments in the chain, respectively.

Table 6. Gaussian Noise Information for the AOP and Pb Data Sets

feature	Gaussian noise standard deviation for the AOP data	Gaussian noise standard deviation for the Pb data
reservoir pressure	0.219	0.229
reservoir temperature	0.371	0.380
API gravity	0.116	0.126
GOR	1.196	1.301
fluid molecular weight	0.472	0.415
oil-N ₂	0.027	0.005
oil-CO ₂	0.018	0.022
oil-H ₂ S	0.025	0.028
oil-C ₁	0.202	0.225
oil-C ₂	0.042	0.042
oil-C ₃	0.024	0.027
oil-i-n-C ₄	0.043	0.045
oil-i-n-C ₅	0.030	0.030
oil-C ₆	0.031	0.030
saturates	0.252	0.252
aromatics	0.164	0.150
resins	0.077	0.080
asphaltenes	0.097	0.104
N ₂	0.047	0.002
CO ₂	0.201	0.125
H ₂ S	0.001	0.037
C ₁	0.156	0.172
C ₂	0.031	0.022
C ₃	0.021	0.014
C ₄₊	0.015	0.015
temperature	0.637	0.634

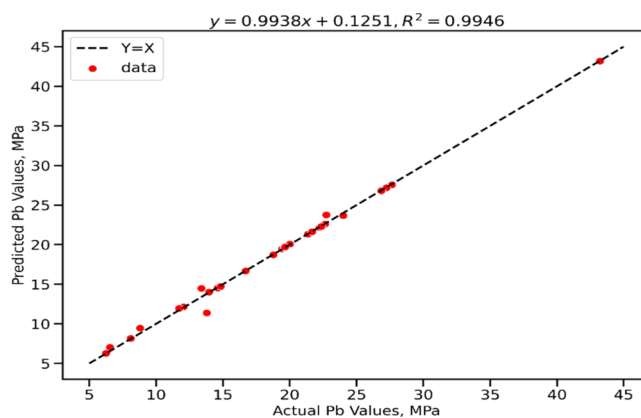
**Figure 7.** Performance of the extra trees model in prediction of the AOP values using the noisy data samples.

3. RESULTS AND DISCUSSION

In this section, the results of the developed models are presented and discussed. Then, the modeling results are compared with the experimental data and the PC-SAFT model.

3.1. Data Statistics. Statistical information for the features and the target variables for AOP and Pb is presented in Tables 2 and 3, respectively.

3.2. Statistical Analysis. Performances of the developed models in predicting AOP and Pb are summarized in Tables 4 and 5. The fitting and evaluation process was performed 100 times for each model to test repeatability of the results. Based on Tables 4 and 5, the ET ($R^2 = 0.793$) and the SVM ($R^2 =$

**Figure 8.** Accuracy of the support vector machine model in prediction of the Pb values using the noisy data samples.

0.988) models are the best predictor of AOP and Pb, respectively. In general, the error analysis performance for the testing results in Tables 4 and 5 shows that the developed AI/ML methods are more reliable in predicting Pb compared to AOP.

3.3. Graphical Analysis. Graphical analysis was carried out to understand performance of the developed AI/ML models. A comparison between the predicted and experimental values of AOP and Pb in the testing phase is presented in Figures 3 and 4. Based on Figures 3 and 4, panels (a)–(d), the ET and the SVM models have the highest accuracy in the estimation of AOP and Pb, respectively.

The residual analysis that presents relative deviation (%) of the predicted AOP and Pb values and the experimental data for the testing data set are presented in Figures 5 and 6. Equation 9 was used to determine the relative deviation as follows:

$$\text{relative deviation (\%)} = \frac{y_{\text{exp}} - y_{\text{pred}}}{y_{\text{exp}}} \times 100 \quad (9)$$

where y_{exp} and y_{pred} are experimental and predicted values, respectively. Based on Figures 5 and 6, the ET model yielded the best performance and the majority of the deviations accumulated between -20% and 20% . In addition, the SVM model provided an accurate efficiency for Pb (deviation between -5% and 5%).

3.4. Reliability Test. According to the graphical and statistical analyses, the SVM and the ET models showed the best performance in estimating Pb and AOP, respectively. However, the model was tested using noisy data from the original data set. A random 20% of the original data set was chosen, and noisy samples were generated by adding Gaussian noise to each feature in the data set. The information related to the noises that are added to each feature is presented in Table 6.

Efficiency of the ET and the SVM models (the best models) in estimating AOP and Pb parameters, respectively, using the noisy data samples is presented in Figures 7 and 8. These figures confirm the accuracy of the developed models in prediction of the target values.

3.5. Comparison between Smart Models and the PC-SAFT. The results from comparing the experiment and AI/ML models in estimating the AOP and Pb with the results of the PC-SAFT model reported by Panuganti et al. are presented in Table 7.¹⁵ Based on the results, the PC-SAFT model has a better performance in predicting AOP during gas injection

Table 7. Compression between the Experimental Data, the PC-SAFT Model, and the AI/ML Models Developed in This Research Work

	AOP						Pb					
	experimental	PC-SAFT	ET	KNN	DT	SVM	experimental	PC-SAFT	ET	KNN	DT	SVM
5% gas injection	3414.94	3414.94	2473.64	1925.92	1806.88	2723.20	1761.71	2003.65	1520.96	1559.37	1483.48	1696.12
	2766.24	2604.95	2005.80	1563.16	1466.70	2207.42	1959.79	2121.08	1691.71	1734.74	1649.98	1886.77
	3397.26	3360.37	2460.89	1916.03	1797.61	2709.14	2155.25	2301.37	1860.19	1907.26	1814.29	2074.90
	2958.90	2776.26	2144.75	1670.90	1567.73	2360.60	2301.37	2447.49	1986.15	2036.43	1937.11	2215.54
15% gas injection	4285.71	4489.80	3101.64	2412.85	2263.51	3415.55	2285.71	2489.80	1972.65	2022.59	1923.95	2200.47
	3714.29	4040.82	2689.54	2093.31	1963.86	2961.21	2775.51	2693.88	2395.86	2455.57	2335.68	2671.90
	4740.74	4740.74	3429.81	2667.3	2502.13	3777.34	2444.44	2481.48	2109.47	2162.90	2057.38	2353.24
	4037.04	3851.85	2922.30	2273.79	2133.11	3217.83	2518.52	2592.59	2173.33	2228.39	2119.65	2424.54
30% gas injection	3703.70	3592.59	2681.90	2087.39	1958.31	2952.79	2888.89	2814.81	2492.59	2555.80	2430.98	2781.03
	9183.67	9183.67	6634.05	5151.79	4832.00	7309.92	3020.41	3020.41	2605.96	2672.06	2541.54	2907.61
	7020.41	6489.80	5073.91	3942.09	3697.59	5589.91	3428.57	3265.31	2957.79	3032.87	288.64	3300.47
	8444.44	8222.22	6100.92	4738.41	4444.35	6722.15	3037.04	3296.30	2620.30	2686.76	2555.52	2923.62
	6185.19	6370.37	4471.55	3475.04	3259.60	4925.82	3555.56	3370.37	3067.26	3145.13	2991.39	3422.70
R ²		0.99	0.79	0.62	0.48	0.76		0.99	0.97	0.94	0.88	0.99
RMSE		1.50	7.51	10.16	11.87	7.99		1.07	1.71	1.84	2.51	0.79

compared to AI/ML models. As mentioned in the previous section, accuracy of the ET (R^2 : 0.79) is higher than that of the SVM (R^2 : 0.76), the KNN (R^2 : 0.62), and the DT (R^2 : 0.48) for AOP. Hence, more input data and experimental points should be used in the AI/ML models for prediction of the AOP to minimize the deviation from actual data. Based on Table 7, the results of the AI/ML models in prediction of Pb were acceptable compared to the PC-SAFT model. The PC-SAFT model (R^2 : 0.99) has similar accuracy to the SVM (R^2 : 0.99) model in prediction of Pb. In addition, the results of the ET (R^2 : 0.97) and KNN (R^2 : 0.94) are reliable for Pb prediction. The amounts of R^2 and RMSE for the AI/ML models presented in Table 7 are based on all data points, not only data from Panuganti et al.¹⁵ Hence, it can be concluded that the SVM, the ET, and the KNN models are a good alternative for the PC-SAFT model for prediction of Pb without using complex mathematical relations. However, in the case of AOP, the results of the PC-SAFT model are better than those of the developed AI/ML models. It is worth mentioning here that in the AI/ML models, data from different studies reported in the literature were used, while in the PC-SAFT model only data from Panuganti et al.¹⁵ were considered. Hence, the accuracy of the PC-SAFT is higher than that of AI/ML models and has a lower probable error. As a result, for prediction of Pb, the AI/ML model is a reliable alternative to the PC-SAFT model and can predict the results successfully without considering complex equations and computational efforts. In addition, the AI/ML model can calculate parameters in various ranges of temperature, while the PC-SAFT model has a poor performance at low temperatures.⁹

Although numerous experimental and theoretical research studies attempted to predict the pressure that asphaltene starts to precipitate during gas injection, there are still several challenges associated with these techniques. Some of the challenges include high computational requirements and efforts, high cost, and the need to solve complex mathematical relationships. It can be concluded here that soft computing methods can assist in overcoming the limitations and estimating target values in a simpler way and with an acceptable error. However, the accuracy of smart methods highly depends on input parameters, which is the main drawback of these methods.

4. CONCLUSIONS

Pressure decline in oil reservoirs during gas injection is a problematic phenomenon that causes asphaltene precipitation and deposition leading to flow assurance issues. Reservoir temperature, reservoir pressure, SARA fraction, API gravity, GOR, fluid molecular weight, monophasic composition, and composition of gas injection were determined by various researchers to be the parameters affecting estimation of the AOP and Pb. Hence, in this research work we attempted to develop four predictive AI/ML-based intelligent models (e.g., ET, DT, SVM, and KNN) to estimate AOP and Pb which are considered as critical parameters in determining the onset of asphaltene precipitation in oil reservoirs. A comprehensive database was collected from experimental research studies reported in the literature for this aim (AOP: 170 data points and Pb: 146 data points). The data set used in the present research work is relatively large and covers all available data from different sources. Missing value imputation, data statistics, and data scaling are three main preparing data methods that we used for preprocessing the database for AI/ML modeling. The following main five conclusions can be drawn from this research work:

- 1- The error analysis and graphical studies showed that the ET (R^2 : 0.79, RMSE: 7.5) and the SVM (R^2 : 0.98, RMSE: 0.76) are the best predictors in estimating AOP and Pb, respectively. In addition, the results of the DT method showed notable deviation from the real data for both data sets. The accuracy of the developed smart models for prediction of the AOP and Pb in the present research work is in the following order:
AOP: ET > SVM > KNN > DT
Pb: SVM > ET > KNN > DT
- 2- The reliability test results using randomly generated noisy data confirmed accuracy of the developed ET (R^2 : 0.998) and the SVM (R^2 : 0.994) models in predicting the AOP and Pb, respectively, compared to the AI/ML models used in the present research work.
- 3- Generally, the accuracy of the PC-SAFT model is more than that of the AI/ML models. However, our results confirm that the AI/ML approach is an acceptable alternative for the PC-SAFT model when we face lack of data and/or complex mathematical equations.

- The error analysis results revealed that the AI/ML model performance in predicting the Pb is more reliable than the AOP.
- The outcomes of the present research work show a promising pathway in estimating the probability of asphaltene precipitation during gas injection using smart computational techniques. The gas injection rate can affect the AOP and Pb. Hence, it would be interesting to consider this parameter in future AI/ML modeling efforts.

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Notes

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NOMENCLATURE

Acronyms

ANN	artificial neural network
AOP	asphaltene onset pressure
CPA	Cubic Plus Association
DT	decision tree
EOS	equation of state
GA	genetic algorithm
GOR	gas–oil ratio
LAOP	lower asphaltene onset pressure
LSSVM	least-squares support vector machine
MELM	multilayer extreme learning machine
OBM	oil-based mud
OFVF	oil formation volume factor
Pb	bubble point pressure
PC-SAFT	Perturbed-chain form statistical associating fluid theory
PR	Peng Robinson
PSO	particle swarm optimization
QCR	quartz crystal resonator
RF	random forest
SAFT	Statistical Associating Fluid Theory
SRK	Soave–Redlich–Kwong
STO	stock tank oil
SVM	support vector machine
SVR	support vector regression
UAOP	upper asphaltene onset pressure

Variables and Parameters

\tilde{a}^{res}	total residual Helmholtz free energy
\tilde{a}^{disp}	force between nonpolar and weakly polar molecules
\tilde{a}^{assoc}	association contribution
\tilde{a}^{hc}	hard-chain contribution
d	sphere diameter
k	Boltzmann constant
m_i	number of the segment in the chain
N	number of molecules
T	temperature
x	molar fraction of chain component

Greek Letters

ϵ	segment energy
$\epsilon_{A,B}$	associate energy
$\kappa_{A,B}$	associate volume
σ	segment diameter

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